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# $Rb_4Mo_{21}Se_{24}$ containing $Mo_{12}$ and $Mo_{15}$ clusters

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Rubidium molybdenum selenide,  $Rb_4Mo_{21}Se_{24}$ , crystallizes in the trigonal space group  $R\overline{3}$ . Its crystal structure consists of a mixture of  $Mo_{12}Se_{14}Se_6$  and  $Mo_{15}Se_{17}Se_6$  cluster units in a 1:2 ratio. Both units are interconnected through Mo–Se bonds. The Rb<sup>+</sup> cations occupy large voids between the different cluster units.

## Comment

The crystal structures of reduced molybdenum chalcogenides are characterized by metal-metal bonds which give rise to discrete clusters of diverse sizes and geometries. Compounds containing such clusters can be classified as two groups. The first group comprises the reduced molybdenum chalcogenides whose crystal structures contain only one type of cluster and is particularly well exemplified by the series of compounds  $M_{n-2}Mo_{3n}X_{3n+2}$  (M = Rb, Cs; X = S, Se or Te; n = 3, 4, 5, 6, 7, 8, 10 and 12) containing Mo<sub>9</sub>, Mo<sub>12</sub>, Mo<sub>15</sub>, Mo<sub>18</sub>, Mo<sub>21</sub>, Mo<sub>24</sub>, Mo<sub>30</sub> and Mo<sub>36</sub> clusters (Gougeon, 1984; Gougeon et al., 1984, 1987, 1988, 1989a,b, 1990; Thomas et al., 1997; Picard, Gougeon & Potel, 1999; Picard, Halet et al., 1999). The second group is based on compounds containing clusters of odd and even nuclearity in equal proportion and is represented by the series of compounds  $Rb_{2n}Mo_9X_{11}Mo_{6n}X_{6n+2}$  (*n* = 1, 2, 3, 4 and 5; Picard, Gougeon & Potel, 1999). Interest in these Mo cluster compounds lies not only in their fascinating structural aspects, but also in their interesting physical properties. Indeed, the sulfides and selenides generally present superconducting or metal-insulator transitions at low temperature. We present here a new structural type, Rb<sub>4</sub>Mo<sub>21</sub>Se<sub>24</sub>, also containing odd and even nuclearity clusters but in the ratio 1:2. Indeed, the crystal structure of Rb<sub>4</sub>Mo<sub>21</sub>Se<sub>24</sub> is made up of Mo<sub>12</sub>Se<sub>14</sub>Se<sub>6</sub> and Mo<sub>15</sub>Se<sub>17</sub>Se<sub>6</sub> cluster units in a 1:2 ratio. Both units share part of their Se atoms according to the connectivity formula  $Mo_{12}Se_8^iSe_{6/2}^{i-a}Se_{6/2}^{i-a}$ ,  $2Mo_{15}Se_{11}^iSe_{6/2}^{i-a}Se_{6/2}^{i-a}$  to form large voids in which the Rb<sup>+</sup> ions reside. The Mo<sub>12</sub>Se<sub>14</sub>Se<sub>6</sub> and Mo<sub>15</sub>Se<sub>17</sub>Se<sub>6</sub> cluster units are similar to those reported previously in Rb<sub>2</sub>Mo<sub>12</sub>Se<sub>14</sub> (Gautier et al., 1998) and Rb<sub>3</sub>Mo<sub>15</sub>Se<sub>17</sub> (Gougeon et al., 1989b), respectively. The Mo cores of both units result from the mono-axial trans-face-sharing of three

and four octahedral Mo<sub>6</sub> clusters, respectively, and can be described alternatively as resulting from the stacking of four and five Mo3 triangles, respectively. The Mo12Se14Se6 unit possesses the same local symmetry ( $\overline{3}$  or  $C_{3i}$ ) and does not show any significant structural differences from that observed in Rb<sub>2</sub>Mo<sub>12</sub>Se<sub>14</sub>. Indeed, the intracluster Mo-Mo and Mo-Se distances are similar within 0.008 Å in both compounds, indicating an identical cationic transfer towards the Mo<sub>12</sub>Se<sub>14</sub>Se<sub>6</sub> units. The symmetry of the Mo<sub>15</sub>Se<sub>17</sub>Se<sub>6</sub> unit is 3 or  $C_3$ , while it is  $\overline{6}$  or  $C_{3h}$  in Rb<sub>3</sub>Mo<sub>15</sub>Se<sub>17</sub>. Despite the lowering of the local symmetry of the Mo<sub>15</sub>Se<sub>17</sub>Se<sub>6</sub> cluster unit due to the loss of the mirror plane, the different Mo-Mo and Mo-Se distances within the  $Mo_{15}Se_{17}Se_6$  unit compare well with those found in Rb<sub>3</sub>Mo<sub>15</sub>Se<sub>17</sub>. Indeed, the greatest deviations observed are 0.005 and 0.01 Å for the Mo-Mo and Mo-Se bonds, respectively. The shortest Mo-Mo distance between the  $Mo_{12}$  and  $Mo_{15}$  clusters is 3.267 (2) Å, and between the  $Mo_{15}$  clusters is 3.280 (3) Å. The latter two distances are again similar to those found in  $Rb_2Mo_{12}Se_{14}$  [3.279 (1) Å] and  $Rb_3Mo_{15}Se_{17}$  [3.268 (2) Å]. The Rb1 atom is surrounded by 11 Se atoms forming a distorted trigonal prism capped on all faces. The other alkali-metal cations are arranged in finite chains along the threefold axis between two consecutive Mo<sub>15</sub>Se<sub>17</sub>Se<sub>6</sub> units in a distorted tri- or tetracapped trigonal prismatic environment of Se atoms. The cations at both ends of the finite chain are in a tetracapped trigonal prismatic arrangement, and the remaining cations are in a tricapped trigonal coordination. The latter environment is similar to that



### Figure 1

Plot showing the atom-numbering scheme and the inter-unit linkage of  $Mo_{12}Se_{14}Se_6$  and  $Mo_{15}Se_{17}Se_6$  cluster units. Displacement ellipsoids are drawn at the 97% probability level.

encountered in the quasi-one-dimensional compounds  $M_2Mo_6X_6$  (Potel, 1981). The Rb—Se distances spread over a wide range [3.508 (2)–3.973 (5) Å]. The shortest distance between adjacent Rb<sup>+</sup> cations is 4.695 (4) Å.



Figure 2 View of Rb<sub>4</sub>Mo<sub>21</sub>Se<sub>24</sub> along [110].

# **Experimental**

Single crystals of  $Rb_4Mo_{21}Se_{24}$  were obtained in an attempt to prepare  $Rb_4Mo_{21}Se_{25}$ . They were prepared from a mixture of  $Rb_2Mo_6Se_6$ ,  $MoSe_2$  and Mo in the proportions 2:6.5:2.5. The ternary compound  $Rb_2Mo_6Se_6$  was synthesized from an ion-exchange reaction of  $In_2Mo_6Se_6$  with RbCl at 1073 K, as described previously. All materials were handled in an argon-filled glove box. The initial mixture (*ca* 5 g) was cold pressed and loaded into a molybdenum crucible, which was sealed under a low argon pressure using an arcwelding system. The temperature was increased at a rate of 300 K h<sup>-1</sup> to 1973 K and held for 6 h, then cooled at a rate of 100 K h<sup>-1</sup> to 1373 K and finally furnace cooled.

## Crystal data

Rh.Mo. Ser	Mo Ka radiation
10213024	WIO Ku Taulation
$M_r = 4251.66$	Cell parameters from 48 971
Trigonal, R3	reflections
a = 9.5078 (2)  Å	$\theta = 1.0-40.3^{\circ}$
c = 87.797(2) Å	$\mu = 28.81 \text{ mm}^{-1}$
$V = 6873.4 (3) \text{ Å}^3$	T = 293 (2) K
Z = 6	Irregular, black
$D_x = 6.163 \text{ Mg m}^{-3}$	$0.13 \times 0.04 \times 0.04 \text{ mm}$

### Data collection

Nonius KappaCCD diffractometer 5050 reflections with  $I > 2\sigma(I)$  $R_{\rm int} = 0.094$  $\varphi$  ( $\kappa = 0$ ) plus additional  $\omega$  scans Absorption correction: empirical  $\theta_{\rm max} = 40.22^\circ$ (SORTAV; Blessing, 1995)  $h = -17 \rightarrow 12$  $k = -15 \rightarrow 17$  $T_{\min} = 0.247, \ T_{\max} = 0.468$ 33 750 measured reflections  $l=-157\rightarrow 158$ 9548 independent reflections Intensity decay: none Refinement Refinement on  $F^2$  $\Delta \rho_{\rm max} = 2.96 \text{ e} \text{ Å}^{-3}$  $R[F^2 > 2\sigma(F^2)] = 0.051$ (near Mo5)  $wR(F^2) = 0.117$  $\Delta \rho_{\rm min} = -2.62 \text{ e } \text{\AA}^{-3}$ S = 0.959(near Mo6) 9548 reflections

9548 reflectionsExtinction correction:149 parametersSHELXL97 $w = 1/[\sigma^2(F_o^2) + (0.0399P)^2]$ Extinction coefficient:where  $P = (F_o^2 + 2F_c^2)/3$ 0.000149 (5) $(\Delta/\sigma)_{max} = 0.003$  $\Delta/\sigma$ 

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *COLLECT*; data reduction: *DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *DIAMOND* (Bergerhoff, 1996); software used to prepare material for publication: *SHELXL*97.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1436). Services for accessing these data are described at the back of the journal.

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